AN INVESTIGATION OF GAS/WATER/ROCK INTERACTIONS & CHEMISTRY

 Develop reservoir or basin scale models that include flow, mass transport, and chemical reaction processes for CO₂ injection and field pilot test sites.

Accomplishments

The facilities to conduct hydrothermal CO₂-water-rock reactions and analyze these complex mixtures have been developed at NETL. Work on the systematic study of the solubility of CO₂ in increasingly complex salt solutions is currently underway.

In addition to construction of a database containing physical and chemical information on over 64,000 brine wells, NETL has added information on the locations of coal-fired power plants and information on seismic activity. A composite map depicting the power plants, saline formations, and seismic potential was constructed. The high-pressure chemistry of CO₂ with brines sampled around the nation has been started. The pertinent reactions have been identified and the effect of temperature, pressure, pH, and other variables determined. Lastly, several simulations of brine field sequestration have been developed. These include development of sophisticated reservoir models as well as reactive transport models.

Benefits

This project will provide useful information in the area of reaction kinetics dealing with carbon dioxide and surrounding minerals and also provide a compiled brine database of some 64,000 brine wells in the United States. By compiling a database of these brines along with power plants and seismic activity in the United States, a more efficient means of storage can take place in optimal locations. Taking nearby power plant emissions and local seismic activity into consideration, researchers and engineers become more informed as to where precautions need to be taken or simply where areas of higher risk are located. Thus, an understanding of the fundamental chemistry associated with the reactions coupled with a detailed brine database provides much needed information and efficiency to the actual sequestration projects. Additionally, by capturing carbon dioxide and sequestering it, harmful emissions into the atmosphere are prevented that may further increase global warming.

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About two thirds of the United States is underlaid by deep saline aquifers that have an estimated CO₂ adsorption capacity of between 320 to 10,000 billion tons. Unfortunately, there are a large number of uncertainties associated with the heterogeneous reactions which may occur between CO₂, the brine, and minerals in the surrounding strata—especially with respect to reaction kinetics. This project focuses on the complex solution and surface chemistry of CO2 in brines in the presence of host rock and the special types of analyses required to study the reaction kinetics. Carbonate mineral formation/dissolution reactions that may be important in geologic sequestration in deep saline aquifers will be identified. The kinetics of CO₂ dissolution in the liquid phase and subsequent substrate-water reactions are slow and poorly understood. Understanding the kinetics of both these types of reactions and the processes controlling them is essential to understanding the conversion of CO₂ into stable carbonate minerals.

A compilation of existing brine data from a variety of sources, and a complete statistical analysis of the brine chemistry and other geological parameters associated with brine aguifers would be a valuable tool for both experimental and modeling studies of CO₂ sequestration in brines. Currently, NETL is developing a brine database that includes temperature, depth, pressure, and a variety of chemical variables (pH, sodium, iron, chloride, bicarbonate, calcium, magnesium, sulfate, and total dissolved solids) on some 64,000 brines taken from the contiguous United States. Sources of these data include those provided by the USGS, searches of geoscience literature, State Geological Surveys and oil and gas producing companies. Additionally, NETL has instituted a limited field program of brine collection throughout the United States. This brine sampling is being done in conjunction with other government agencies and oil and gas companies.

Primary Project Goal

The ultimate objective of the work being performed jointly at NETL and the United States Geological Survey is to carry out an experimental study to assess the role of the chemistry of formation water in CO₂ solubility. Then the role of rock mineralogy in determining the potential for CO₂ sequestration through geochemical reactions will be assessed.

Objectives

- Investigate kinetics of CO₂ dissolution in brines at temperatures and pressures appropriate for deep saline aquifer carbon dioxide sequestration.
- Improve the understanding of the processes by which mineral carbonates are formed and study the reactivity of various mineral substrates involved in these processes.
- Assess and collect both brines and surrounding geologic strata in selected brine formations in the conterminous United States in order to determine their potential to sequester CO₂ from fossil fuel fired power plants.

CONTACT POINTS

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Background

National Energy Technology Laboratory United States Geological Survey Parsons Power

Battelle Memorial Institute University of Pittsburgh California University of Pennsylvania University of Texas Case Western Reserve University

PRIMARY PARTNERS

DOE FUNDING PROFILE

Prior FY's \$682,000 FY2002 \$817,000 TBA Future FY

TOTAL ESTIMATED COST

DOE

\$1,499,000

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